**USE OF MICROBIAL FUEL CELL IN GENERATING ELECTRICITY FROM WASTE MATERIAL**

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**ABSTRACT**

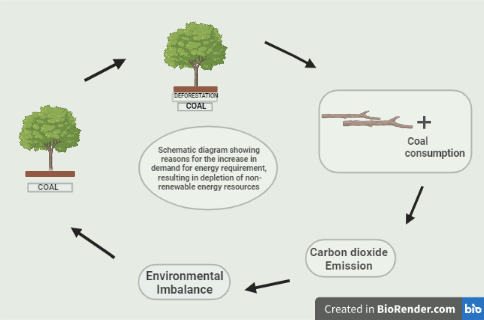
The demand for power is quite significant on a global scale. Microbial Fuel Cell (MFC) Technology may be used to lessen dependance upon fossil-fuels and to provide alternative sustainable energy sources. MFC Technology uses microorganisms to produce power using the organic matter found in the environment. MFC is a biofuel cell, that generates electricity by converting organic material into electricity. Due to its ability to use waste-water as a substrate and to not require a metal catalyst, it can be taken into consideration as a more sustainable alternative for traditional fuel cells. Waste material are first transformed to chemical energy and then, after being treated to the desired level to electrical energy. An anode, cathode and a separation membrane are the basic MFC components. MFC technology is likely to become a more environment friendly fuel cell alternative. Despite being viewed as a promising technology, MFC is not yet commercially viable for usage on a large scale due to its poor current generation per unit cost and high internal resistance. More study should be conducted on the creation of more efficient electrode materials and the development of resilient microorganisms as biocatalysts in order to boost the viability of MFC technology.

**Keywords:** Microbial fuel Cell (MFC), Waste, Waste-water, Microorganisms, Electricity, MFC Technology

1. **INTRODUCTION**

Human population on earth is already on the verge of reaching 8 billion and considering the current scenario that shows rapid growth in population, by the second half of the twenty-first century, it is predicted that this number will increase even more and level out. Because of the ambitious social and economic aims, it has become important to worry about the sustainable use and management of natural resources. (A. S. Vishwanathan 2021) In the contemporary world, energy plays a very crucial role. Demand for energy is rising rapidly with increasing population and urbanization. Primarily fossil-fuels continue to be a main source of energy, and this causes fossil fuel depletion in their reserves. Combustion of fossil-fuels also produces large amount of carbon dioxide (CO2), which raises the atmospheric concentration of greenhouse gases, ultimately leading to global warming. Over the past ten years, alternative energy sources that are more cost-effective, renewable, and environment friendly have come as a major point of focus for the researchers. (Rustiana Yuliasni et.al.,2021)

Considering the spontaneous rise in the energy demand, researchers have come forward to develop a solution for renewable alternative sources of energy such as MFC Technology. It is a promising technique for energy production that falls under the category of secondary energy production. MFC has the concurrent capacity to treat waste-water and at the same time produce electricity. Additionally, MFCs are seen as an environmentally benign alternative source to fossil-fuels. MFC appear to be the answer to our twin issues of rising pollution and declining reliance on fossil-fuels. It is a very good alternative energy source that is both highly efficient and economically priced. (A. Sam Sushmitha, Dr. G.L. Sathyamoorthy, 2020) In 1917, researchers used *Escherichia coli* to develop the idea of generating electricity using microorganisms. Since then and till date, MFC technology, which was first developed in the 1990s, concentrated more on increasing the output of power generation, reducing the costs of operation, and making this technology more useful and sustainable. (Rustiana Yuliasnet.al., 2021) MFCs can use microorganisms as catalysts to oxidise organic matters (mud, food wastes, vegetables and fruits wastes, grass fragments and leaves of plants) and inorganic substances (non-carbon materials, sulphur compounds, etc.) to generate electricity. Researchers also demonstrated that hydrogen could be created efficiently by MFC, which is used for supplying electricity to the system and for the purification of water. Since hydrogen remains persistent during both the process of combustion and the electrochemical process for energy combustion, it can reduce or eliminate carbon emissions. (Kasirajan Kasipandian et.al.,2020)



**Figure 1. depletion of non-renewable energy resources due to increasing energy demand**

1. **AIM OF THE REVIEW**

A Microbial Fuel Cell (MFC) is a bio-electrochemical device that has the ability to generate electricity by the use of electrons obtained by the anaerobic oxidation of substrates. This review mainly focuses on the use of MFC technology using waste material as an alternative source for fossil-fuels to fulfil the need of rising energy demands due to the rapidly increasing population.

1. **REVIEW OF LITERATURE**

MFC is a system where microbes transform chemical energy from the oxidation of organic and inorganic chemicals into ATP through a series of processes where electrons are transported to a terminal electron acceptor producing electrical current. Anode and cathode components, separated by cationic membrane, makes a conventional MFC. [4] Figure 2 shows the basic working principle of MFC. In an MFC, an ion exchange membrane may or may not be used to separate the anodic and cathodic chambers. Live microorganisms (electroactive microorganisms) oxidise substrates while in a planktonic state or by developing biofilms in the anode-chamber, which leads to the production of electrons and protons, and other metabolites as by-products. (Soumya Pandit, 2018) The anode electrons go through the external circuit to the cathode and provide electron acceptors in cathode region because of the potential difference between the cathode and the anode. (Shuyi Zhou, 2022) On the other side, the protons diffuse to the cathode or percolate through the ion exchange membrane, where they are reduced by the incoming electrons, completing the circuit. Electric current is produced by the electron flow in external load. Water is produced as a by-product as protons gets reduced in the presence of oxygen at anode, making the process environmentally benign. MFCs come in both big (litre-scale) and tiny sizes (microlitre-scale or millilitre-scale). (Soumya Pandit,2018)

Using glucose as a substrate, following are the reactions:

Anode: C6H12O6+ 6 H2O → 6 CO2 + 24 H + 24 e− (1)

Cathode: 4 e− + 4 H+ + O2 → 2 H2O (2)

Overall: C6H12O6 + 6 H2O + 6 O2 → 6 CO2+ 12 H2O (3) (Rustiana Yuliasniet.al.,2021)

**3.3. ESSENTIAL MFC COMPONENTS**

**Table 1 shows the list of MFC components**.

|  |  |  |
| --- | --- | --- |
| COMPONENTS | MATERIALS | REMARKS |
| Anode | Graphite, carbon paper, graphite felt, carbon-cloth, Pt black, Pt, etc. | Essential |
| Cathode | Graphite, carbon paper, graphite felt, carbon-cloth, Pt black, Pt, etc. | Essential |
| Anodic Chamber | polycarbonate, Glass, plexigals | Essential |
| Cathodic Chamber | polycarbonate, Glass, plexigals | Optional |
| Proton-exchange system | Nafion, Polyethylene, Ultrex | Essential |
| Electrode catalyst | Pt, Pt black, polyaniline, etc. | Optional |

**3.3.1. Electrodes**

The performance of MFCs in terms of bacterial adhesion, electron-transfer, and electro-chemical effectiveness depends significantly on the choice of suitable electrode material. Despite the differences between anode and cathode, they should both generally have the following qualities: surface-area, porosity, durability, cost, electrical conductivity, accessibility, and stability

**Anode**: Owing to their high electrical conduction, specific surface-area, bio-compatibility, chemical stability, and affordability, graphite rods are commonly used as anodes in MFCs.

**Cathode**: The characteristics that are must for a strong MFC cathode are high mechanical strength, catalytic property, high ionic and electronic conduction. The best cathode is made of graphite rods, however due to its low catalytic activity, an extra catalyst must be added to speed up the reduction process. (Mrs.Xma R.Pote, et.al.,2020)

* + 1. **Proton Exchange Membrane (PEM)**

Electro-neutrality between two chambers in MFC is a vital prerequisite for the PEM’s effective operation, which is made possible by the transport of protons across the membrane. PEM are a vital part of the MFC-assembly as they aid in separating the anode and cathode chambers and make it easy for protons to move from anode to cathode, where they are required to continue the electric current. The standard PEM has desirable attributes such as Cost-effectiveness, improved proton conductivity, good segregational qualities, higher mechanical strength, heat resistance and chemical resistance, and electronic resistance etc (Akshay D. Tharali, et. al.,2016)

**3.3.3. Substrates**

Waste-water is an example of organic material that bacteria use as an energy source (Al Saned AJ, et.al.,2021)

|  |  |  |
| --- | --- | --- |
| WASTES USED AS SUBSRATES | MICROBIAL SPECIES | REFERENCE |
| Food wastes | *Proteobacteria, Bacteroidetes* | [10] |
| Dairy manure | *Clostridium, Pseudomonas* | Zhang et.al. (2012) |
| Swine manure | *Clostridium* | Vilajeliu-Pons et.al. (2015) |
| Sewage sludge | *Lactobacillus, Flavobacterium* | Xiao et.al. (2011) |
| Food waste leachate | *Proteobacteria, Acidobacteria* | Li et.al. (2013) |
| Powder orange peel waste | *Chloroflex* | Miran et al. (2016) |
| Aerobic sludge | *Nitrospirae, Chlorobi* | [7] |
| Anaerobic sludge | *Arcobacter, Geobacter* | Wang et.al. (2011) |
| Waste-water | *Shewanella , mixed cultures* | Vinay Sharma, Patit Paban Kundu (2010) |

Continuous power generation from domestic waste-water is possible. In a single-chambered MFC, it can be seen how greatest power density is produced using swine waste-water as a substrate. Additionally, oil effluent can be used to create bio-electricity. Waste-sludge has also been proved to be a successful substrate for production of hydrogen and bio-electricity together. Microbes isolated from the high Andean region were grown on vegetable and fruit wastes in a single-chambered MFC. Table 2 shows various substrates and their specific microbes used in MFCs.

Food waste leachate from bio-hydrogen fermentation can be used as a feasible substrate for improved power generation. Simple substrates like glucose, propionate, acetate and butyrate have been used in MFCs as substrates. Acetate > butyrate > propionate were the substrates for which the power density was assessed. This is especially crucial since the acidogenic decomposition of organic wastes results in a variety of volatile fatty-acids as per their affinity for microorganisms, for the production of electricity. (Akshay D. Tharali, et. Al.,2016)

**3.3.4. MFC Specific Microorganisms**

MFC were developed with the goal of producing clean electricity from organic wastes. The most prevalent biocatalysts or microorganisms employed in MFCs are *Shewanella, Proteobactor,* and *Pseudomonas* groups. Mixed cultures are most frequently employed in MFCs that utilise waste water. (Vinay Sharma, Patit Paban Kundu, 2010). Firstly, it’s essential to understand some of the major and in depth operations of the bacteria in order to comprehend the basic operations of the MFC. (De Juan et al., 2015)

Previously, it was believed that just a small number of microbes could be employed to generate electricity in the MFC Technology.

In the recent works, it was discovered that MFCs can use the majority of microorganisms. Many microorganisms are capable of transferring electrons to anode that are produced through the metabolism of organic and/or inorganic materials. (Yibrah Tekle and Addisu Demeke, 2015)

Microorganisms with capacity to transmit electrons efficiently and produce electricity in the anode are specifically taken into consideration by the researchers. Such microbes are called Exo-electrogens. Exo refers to “exocellular” and “electrogens” as per on their ability to directly transfer electrons to a chemical and/or material that is not an immediate electron acceptor (De Juan et al., 2015).These microorganisms are plentiful in marine sediment, fresh water sediment, soil, waste-water, and activated sludge. (Yibrah Tekle and Addisu Demeke, 2015 )

**3.4. DESIGN AND OPERATION**

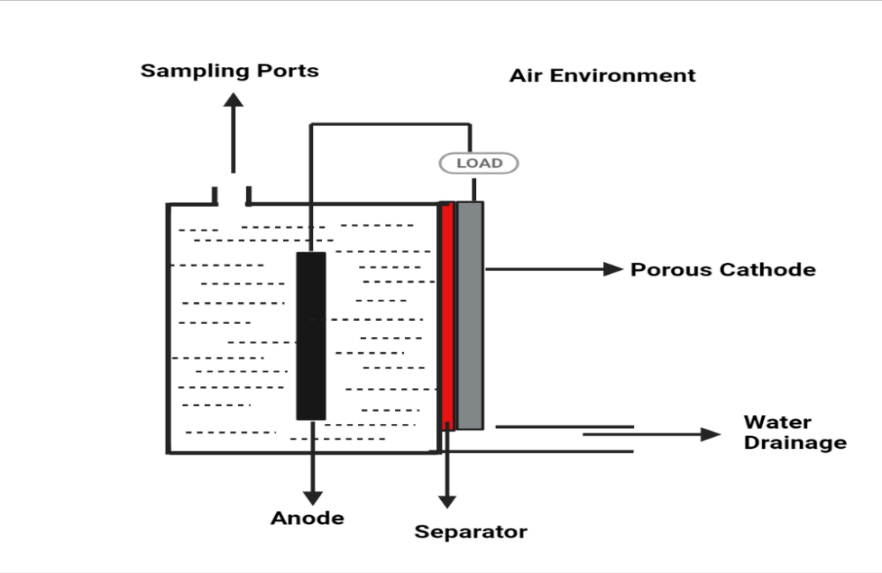
The formation of Microbial Fuel Cells depends on few key components: Wirings, salt bridges, and electrodes. In a PEM power device or fuel cell, the proton exchange membrane replaces the salt bridge ( Ieropoulos I et al., 2012). Though the costs rise up, but the handling and power generation both improve and expand. In addition, making the complete system more portable and efficient. Fuel cells can be divided into two classes based on the number of chambers/compartments they have:

**1.Single-Chambered MFC**  **2. Dual-Chambered MFC**

Additionally, there is the stacked microbial fuel cell (Jumma S, et al., 2016)

**3.4.1. Single-Chambered MFC**

Single-chambered MFCs are available in a variety of designs and can be built in various methods. (Jia J et al., 2013) They are cost-effective and simple in design. (Atkar AB, et al.,2017)



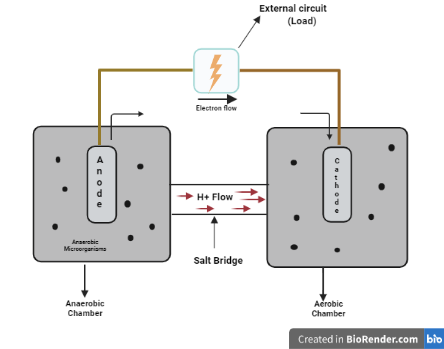
**Figure 3. Single-chamber MFC: anode and cathode electrode are fixed in same chamber connected by external wire**

In this design, anode and cathode are not separated into different compartments/chambers (Jia J et al., 2013). They lack a distinct cathode compartment and have simple anode compartment that may or may not have PEMs, (Figure 3) (Jumma S, et al., 2016). One side of the cathode chamber is made up of porous cathodes, which consume airborne oxygen and permit protons to diffuse through them (Jia J et al., 2013) . Air cathode can be made up of materials such as graphite, carbon paper, fibres. (Atkar AB, et al.,2017)

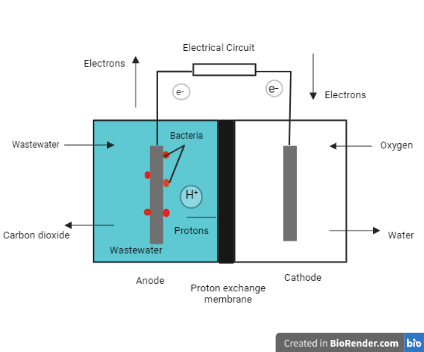
Because they are easier to scale-up in comparison to the double-chambered fuel cells, they have are an extensive subject of research. The cathodes are either porous-carbon electrodes or PEM-bonded with flexible carbon-cloth electrodes. Electrolytes are frequently poured steadily into graphite covered cathodes, which acts as catholytes to keep membrane and cathode from drying. In such single-chambered MFCs, better fluid management is a crucial concern (Jumma S, et al., 2016).

**3.4.2. Dual-Chambered MFC**

Dual-chamber MFCs, also known as two compartment MFCs, have two chambers joined by a salt bridge or PEM (as shown in Figure 4 and 5) which allows protons to go in the direction of the cathode chamber. It may prevent oxygen from diffusing into the anode-chamber. Due to their intricate construction, dual chamber MFCs are frequently utilised in batch mode and for laboratory research. Scaling up is quite challenging in such designs of fuel cells. (Atkar AB, et al.,2017)



**Figure 4. Dual-chamber MFC separated by salt bridge**



**Figure 5. Dual-chamber MFCs separated by PEM**

A variety of materials, including plastic and stainless-steel with coating, can be employed in the design and construction of Dual chambered MFC. The MFC design in Figure 4 is known as the H type as per of its shape. Anode-chamber and cathode chamber are the names of the two chambers. The electrodes can then be installed in each of the two chambers. Graphite or carbon can be used as the electrode’s material. Carbon brush or carbon-clothing can be used as an electrode. H-shaped frameworks are used for fundamental parameter analysis, such as investigating force generation using novel materials or the microbial communities types that rise during the degradation of specific mixes, although they often provide low power densities. The distance between the surfaces of the cathode and the anode, as well as the membrane’s surface, have an impact on how much power is produced in these structures.

High internal-resistance and cathode-based losses typically place restrictions on the power density P supplied by these structures. It is advantageous to compare the power generated by various frameworks based on of comparably sized anodes, cathodes, and membrane wherever possible. Because a good electron acceptor is available at high fixation, using ferricyanide electron acceptor in cathode chamber upsurges the force. Ferricyanide compared to an oxygen disintegrating cathode with a Pt catalyst, the force was increased by 1.5 to 1.8 times. (Jia J et al., 2013)

PEM or salt bridges serve primarily as a proton transfer medium for completing the circuit, as indicated in the Figure 4 and 5. In addition to finishing the reaction, this keeps the anode from getting in touch with oxygen/any other oxidizers. They operate in batches, have a greater power output, and could provide power in very difficult conditions.

It might be scaled up to treat vast volumes of waste-water and other carbon sources. They essentially fall into the single- chambered and double-chamber categories for microbial fuel cells. They lack mediators, occasionally lack membranes, and can be utilised to produce electricity on a huge scale from trash. (Jumma S, et al., 2016)

**3.4.3. Stacked MFCs**

Stack MFCs are MFCs that are coupled in series or parallel to increase output voltage or current. (Atkar AB, et al.,2017) This kind of development doesn’t affect each cell’s individual Coulombic proficiency, but when combined, it increases the overall battery’s yield to be comparable to conventional power sources, as shown in figure 6. (Jia J et al., 2013)



**Figure 6: Stacked type MFC**

There are two different kinds of stack MFCs: one type is the parallel type, and the other is the series type. When operated at same volumetric flow-rate, the parallel connection is six times more efficient than the series connection. The parallel type of stack MFCs have a faster rate of chemical reaction than the series type of stack MFCs. Consequently, parallel-type MFCs are employed to eliminate waste-water’s COD (Chemical Oxygen Demand), which can deliver electrons to the anode directly. (Atkar AB, et al.,2017)

**3.4.5. Up-flow MFC**

The tubular-shaped MFC is composed of layers of glass beads, a glass wall separator, a cathode, and an anode (top and bottom). Food is delivered from below the anode, crosses the cathode, and then travels upward. The diffusion barrier positioned between electrodes creates the gradient necessary for MFCs to function properly. [2] They essentially fall into the single- and double-chamber categories for MFCs. They lack mediators, occasionally lack membranes, and can be utilised to produce electricity on a huge scale from trash. (Jumma S, et al., 2016)

**3.4.6. Other Designs**

Most MFC researches use the basic pre-existing designs, and lesser attention is given in creating new designs that can address the short-comings of the present designs. A flat-plate MFC (FPMFC) was announced in 2004 to decrease one barrier, that is the ohmic-resistance brought on a wider inter-electrode spacing. This style of design is typically employed in the chemical fuel cells, where it also yields greater power generation than the earlier designs. The anode and the cathode in the FPMFC were constructed from the flat-plates, each having a projected surface area of 225 cm2. (Min and Logan, 2004)

Between the two plates was a Nafion membrane. With residential waste-water as a substrate, this reactor generated a power density of 56 mW/m2, and 58% COD was attained in the study. Although the reactor produced lesser power than the other designs, such as the cube reactor, it was also utilised to generate power from different substrates, such as glucose, acetate, starch, etc. It might be because the electrodes are attached too tightly together, allowing oxygen to get through the membrane to microorganism in anode-chamber and hamper the microbes ability to flourish. (Ravinder Kumar, et.al.,2017)

**3.5. APPLICATIONS OF MFC:**

**3.5.1. Electricity generation**

It is clear that the majority of research on MFCs is done for the purpose of producing energy, which is the foremost application of this technology.

**Table 3 shows Performance of MFCs for Bio-electricity Generation.**

|  |  |  |
| --- | --- | --- |
| TYPES OF MFC | SUBSTRATE | POWER DENSITY |
| Single Chambered MFC | Glucose | 68 mW/m2 |
| Acetic-acid | 835 mW/m2 |
| Ethanol | 820mW/m2 |
| Domestic waste water | 114 mW/m2 |
| Double Chambered MFC | Glucose | 855 mW/m2 |
| Acetate | 1926 mW/m2 |
| Acetate | 1.9 Mw/m2 |
| Cellulose | 188 mW/m2 |
| Waste-water | 2485 mW/m2 |

The microorganisms in the MFC's anode-chamber oxidizes the substrate to produce protons and electrons, which are then transferred to the cathode via electrical connection and PEM, respectively. To detect the voltage and calculate the power using Ohm’s law, the two chambers of the MFC can be connected electrically to a multi-meter and an external resistor box. To increase coulombic effectiveness and then the MFCs’ power production, it is critical to use substrates that can totally oxidize into electrons in the MFCs. According to one study, *Geobacter sulfurreducens* may completely convert acetate into electrons and protons (Ravinder Kumar, et.al.,2017)

To increase outputs, electrodes are changed with metal catalyst, nanoparticles, and chemicals. Additionally, the cathode is altered to switch out the expensive platinum catalyst for a less expensive one with equivalent qualities. According to the study, these adjustments reduce the system’s internal resistance and startup time.To improve the output performance, the anode is modified using nitrogen-doped electrodes, heat-treated electrodes, gold nanoparticles, graphene, and carbon nanotubes, for example, CNT-gold-titanium nano-composites improve MFC performance.

Nitrogen doped carbon nano-particles placed on carbon-cloth electrodes are used to support the Extrac-ellular Electron Transfer (EET) mechanism. The Dual-chambered MFC is inoculated with *Shewanella oneidensis* MR-1. By absorbing the flavins the organism secretes, it boosts the rate of electron transmission and power density. According to studies, the anode is dusted with CNT powder, which promotes the growth of the *G. sulfurreducens* biofilm and lowers internal resistance and start-up time. Bacteria are more likely to stick to the electrode because of the faster start-up time. In comparison to the simple carbon-cloth anode, the carbon nano-tube anode in the dual-chambered MFC boosts the density of power up to four times. It has been hypothesised that *Shewanella’s* outer membrane c-type cytochromes exhibit a strong affinity for ferric oxide. The addition of iron oxide to electrodes promotes the EET mechanism and the metabolism of the biofilm along with promoting bacterial growth (Quratulain Maqsood,et.al., 2021).

**3.5.2. Waste water treatment:**

The ability of the MFCs to treat various industrial, urban, or home waste-waters has been potentially observed. Table 4 provides several illustrations of MFC performance for waste-water treatment. Although MFCs are unable to entirely treat the extremely toxic waste-waters, they can significantly lower the COD of the waste-waters to fulfil discharge requirements before being released in the environment. The MFCs have shown to remove up to 98% of COD from waste-water. As an alternative, waste-waters rich in organic substances (such as proteins, minerals, fatty-acids, lipids and carbohydrates) serve as the substrate for microbial metabolism, which in turn generates electrons and protons.

Moreover, inoculum can be found in waste-waters. Before and after the MFC operation, the treatment effectiveness of the MFCs can be assessed using the standard waste-water treatment assays (COD, Biological Oxygen Demand (BOD), total solids, and nitrogen removal). By running the MFCs under ideal conditions, such as mesophilic temperatures, can also increase the COD removal in MFCs. Additionally, the MFC’s fed-batch mode operation is beneficial for achieving high COD removal rates. The coulombic efficiency got in these cases is quite low, varying from 10% to 30%. Typically, MFC studies operated for waste-water treatment are tied with power generation (Ravinder Kumar, et.al., 2017).

|  |  |  |
| --- | --- | --- |
| TYPES OF MFC | WASTE-WATER/HEAVY  METALS | %COD  REMOVAL |
| Single Chambered MFC | Olive mill waste-water | 65 |
| Biodiesel Wastes | 90 |
| Chromium (VI) | 99 |
| Azo dye Congo Red | 98 |
| Cadmium | 90 |
| Double Chambered MFC | Domestic waste-water | 88 |
| Chemical waste-water | 63 |
| Real urban waste-water | 70 |
| Food waste leachate | 85 |
| Cyanide | 88 |

Additionally, several pre-treatment techniques can improve MFC performance. For instance, autoclaving kills the methanogens that use the organic components of waste-water to produce methane. Autoclaving was found to boost power density by 5%. Sonication, on the other hand, was employed to treat untreated waste-water. It boosted COD removal by 5% and power density by 16%. Furthermore, using a stirring approach, COD can be eliminated from waste-water. These pre-treatment methods boost energy production, but they cannot be scaled up. (Quratulain Maqsood,et.al. 2021)

The use of MFC technology for waste-water biotreatment has shown excellently convert organic matter in waste-water to electricity with decrease in COD and BOD of 40-90%. (Oji, et. Al.,2012) Waste-water with a high COD value increases the power density of MFC. However, when the substrate concentration is high, it results in electrode fouling, which restricts flow and causes an accumulation of salts and precipitates. Furthermore, the cathode has less proton availability. Because of this clogging, waste-water dilutions are utilised to control how well the MFCs work.

**3.5.3. Bio-sensors**

MFC technology is used as a biosensor to detect pollutants in water, in addition to producing power and treating waste-water. MFC is designated as a BOD sensor due to the linear relationship between the coulombic yield of the material and waste-water strength. The MFC-based biosensor is superior to traditional bio-sensors in some ways. Because they don’t require a transducer, that is typically utilised in conventional bio-sensors, these bio-sensors are relatively less expensive. Additionally, they can run for a very long time i.e. almost 5 years without any maintenance. MFC-based bio-sensors are therefore more stable and reliable. Numerous studies have demonstrated that vast BOD ranges (low/high) can be monitored in MFC-based bio-sensors on the basis of linear correlation. (Ravinder Kumar, et.al., 2017 & Oji, et. Al.,2012)

The detection of organic matter in waste-water to examine the performance of MFC as bio-sensors. They made two MFC using two different Proton exchange membranes—MFC1 using Nafion and MFC2 using inexpensive clayware—and they investigated how they affected the performance of the MFC as a bio-sensor. (Sumaraj et al.)

The researchers found that MFC1 performs better and responds more quickly to low COD concentrations between 22 mg L-1 and 51 mg L-1. MFC2 can detect concentrations up to 212 mg L-1, but only as high as 67 mg L-1. Both of the MFCs had different reaction times. While MFC2 perceived the concentration at 310 min, MFC1 responded swiftly with sensing times of 210 min and 120 min. They suggested that the PEM employed is the source of the variance in findings by both MFC. Proton conductivity (PC) and membrane thickness are responsible for the change.

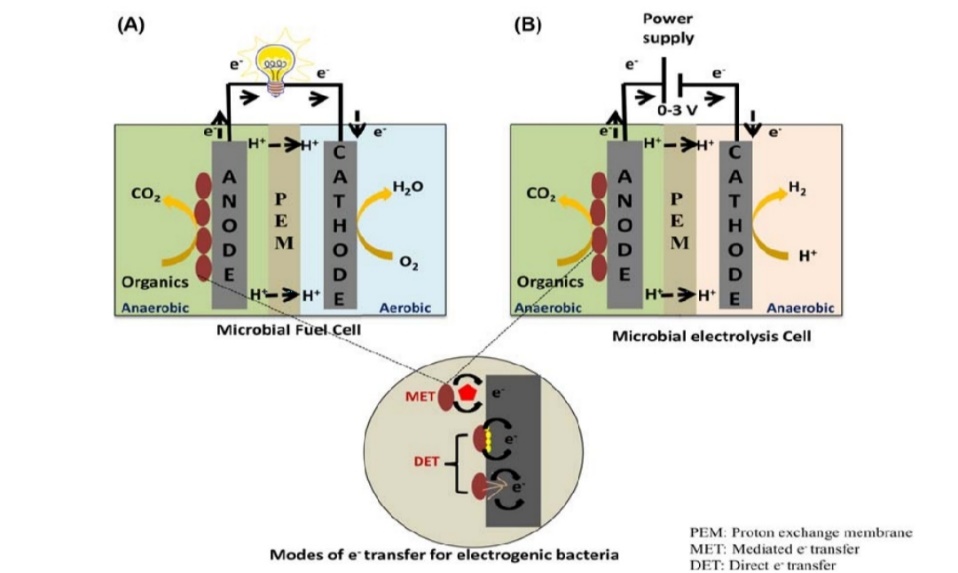
While Clayware (CWPEM) is 17 times thicker than NPEM, Nafion has 40 times more PC than CWPEM. They came to the conclusion that, while taking into account the device’s economics, low-cost clayware PEM might be employed in MFC for BOD monitoring. Although Nafion performs better, it costs 400 cm2 more than clayware, which costs 0.4 cm2. (Abhishek J.et.al., 2019)

**3.5.4. Biohydrogen**

A microbial electrolysis cell (MEC) can be modified from a normal double-chamber MFC to produce hydrogen. The fundamental workings of a MEC remain essentially the same; however, electric current is now supplied to the cathodic chamber. The anode and the cathode are other two chambers that make up MEC. Like MFC, an ion exchange membrane divides the MEC’s two chambers.

Exo-electrogens metabolise the substrate and generate electrons and protons in the anode-chamber. In MFCs, the protons are transported to the cathode similarly. At the cathode, however, a reaction between protons and electrons that would make hydrogen is thermodynamically impossible. At the cathode, electric current is applied to cause this reaction.

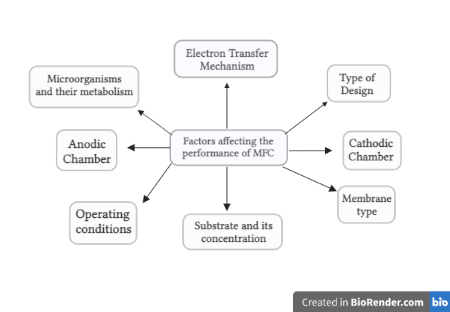
Most of the time, >0.3 V is sufficient to meet the electrical need. The MFCs can readily provide such low voltages. In order to meet the electrical demand, the MFCs used to create power can be paired with MEC. As shown in Figure 9. It is simple to store the hydrogen created by the MEC, which can then be utilised to generate energy. (Ravinder Kumar, et.al., 2017 & Jhansi L. Varanasi et.al., 2019)



**Figure 7. (A) MFC (B) MEC**

**(Jhansi L. Varanasi et.al., 2019)**

**3.6. FACTORS AFFECTING PERFORMANCE OF MFC**



**Figure 8. Factors Affecting MFC’s Performance**

**3.6.1. Electron transfer mechanism**

The electron mediators/ Shuttles must be used to convey the created electrons from the anodic chamber to anode. Some microorganisms, like *Saccharomyces species* and *E. coli strains* used in MFCs, have an outer non-conductive lipid membrane layer that contains peptidoglycans and lipopolysaccharides which slows down the direct electron transfer to anode. The electron mediators are typically coloured substances, like humic acid, thionine, methylene blue, neutral red, or methyl viologen (Shanmuganathan. P et.al.,2018).

The mediators should have the ability to penetrate the bacterial membrane and contact the reductive species inside the bacterium, where they get reduced in microbial metabolism. The mediator’s redox potential should match that of the reductive metabolite. The mediator must not interact with any other metabolites produced by the bacteria. Reduced mediators must be easily transported from the cell to anode, and can be oxidised there. Fast electro-chemical kinetics are required for oxidation of the reduced mediator process at the electrode surface. (Chetan Pandit, et.al.,2022)

**3.6.2. Microbial metabolism and cell potential**

The primary factor that determines the cell potential is metabolic route of the microorganism and consequent potential of the anode. In MFCs, the rate limiting phase is the bacterial catabolism. The oxidation of organic substances delivers energy for the hetero-trophic organisms. The respiratory chain and fermentation are the two key metabolic pathways which takes place in the anode chamber due to the participation of exogenous oxidants, or external terminal electron acceptors.

The potential differences between cathode and anode govern the electrical potential of the MFC. (Shanmuganathan. P,et.al.,2018). Through the respiratory chain, the substrate is oxidised, releasing electrons that go to an externally reachable terminal electron acceptor through a redox cascade. An organism gains extra energy, the higher the positive redox potential of terminal electron acceptor with a given substrate or a electron donor. (Chetan Pandit, et.al.,2022)

**3.6.3. Substrate concentration**

Substrates used in MFCs come in a wide range from straightforward chemicals to intricate organic combinations. Pure substrates like acetate, butyrate, glucose, cellulose, cysteine, lactate, proteins, glycine, and glycerol have occasionally been employed. (Shanmuganathan. P, et.al.,2018)

Waste-water is one of the many substrates that MFCs may process and is a medium with many resources. Several studies on the direct generation of electricity from complex organic waste-water, including municipal, dairy, slaughter house, swine, molasses waste-water, sulfide-rich waste-water, refinery, brewery, winery, rice mill, tannery, cassava mill, chemical waste-water, landfill leachate, azo dye, food waste leachate, and solid substrates like rice straw have been published. MFCs can also simultaneously extract nitrate and sulphide from sulfate-sulfide-rich waste-waters and synthetic effluent. (Chetan Pandit, et.al.,2022) MFC should be operated with the highest level of concentration possible to generate the most electricity possible. (Maksudur R. Khan,et.al.,2013)

**3.6.4. Anode**

The electrode at which electrons leave the cell and where oxidation takes place is called as the anode.

The crucial properties of anode material are conductivity, non-corrosive manner, biocompatibility, stability, and surface area. Additionally, the MFC’s architecture and electrode fabrication process have an impact on how well it performs. Despite having strong conductivity, bacteria cannot tolerate copper, making it an unsuitable material for an anode. Although it has a lower conductivity for moving electrons, carbon is a better electrode material for adhesion of bacteria. It comes in carbon felt, paper, fabric, foam, and fibre forms. Anode materials have a considerable impact on the performance and price of MFCs, which contributes to their effectiveness. (Chetan Pandit, et.al.,2022)

Excellent electrical conductivity, a sizable surface area, and good bio-compatibility for bacterial colonisation are ideal characteristics for an anode in an MFC. The microorganisms in the anode-chamber, which has anaerobic conditions, produce gas that should quickly exit the system to prevent the anode-chamber from building up pressure. Therefore, the requirement for a high bed void fraction is crucial. Another element is the device that directly transfers electrons from the anode to the bacterium, which is what produces the majority of the power. Therefore, increasing the anode’s physical contact with bacteria is beneficial for MFCs’ ability to generate power. (Shanmuganathan. P et.al.,2018)

**3.6.5. Cathode**

The biggest obstacle to an MFC is the cathode design. At the cathode contact, the oxygen reduction reaction (ORR) depletes oxygen and produces either water or hydrogen peroxide. Due to the functioning MFC’s neutral pH and ambient operating temperature, its ORR is inhibited in comparison to chemical fuel cells. The cathode materials, which must have a high redox potential, have a significant impact on the performance of an MFC. Carbon-cloth, Graphite, and commercially available carbon paper are the most frequently utilised materials for cathodes. However, unless metal catalysts are utilised, it is challenging to achieve high cathodic potentials. (Shanmuganathan. P et.al.,2018 & Chetan Pandit et.al.,2022)

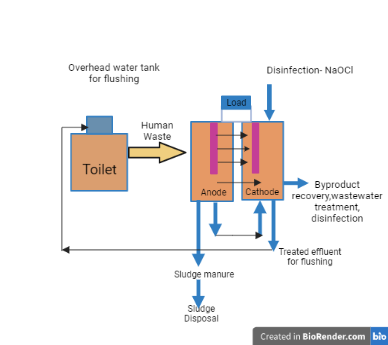
**3.6.6. Operating conditions**

The most crucial elements for bacterial growth that impact MFC effectiveness are temperature, ionic strength, pH, and salinity. The general reaction environment for MFCs is mild, including room temperature, normal pressure, and neutral pH. (Shanmuganathan. P et.al.,2018 & Chetan Pandit et.al.,2022)

**3.7. RECENT ADVANCEMENT**

**Integrating Human Waste In MFC Technology**

The development of MFC technology towards commercialisation has been eased by a number of scientific and technological advances more than 100 years of research. Given the current level of MFC research, the low power output and associated high manufacturing costs of this technology restrict its use in the field. An appropriate and practical approach for long term sanitation to address the inadequate treatment efficacy of the CST system now in use may be field deployment of MFC. Additionally, it is possible to think of human waste (excreta and urine) as a sustainable substrate which can be used in MFC to capture energy during effective waste-water treatment along with recovering by-products (Figure 9) and promoting the reuse of treated water.



**Figure 9. Integration of Human waste in MFC technology**

Such an environmentally benign bio-electro-chemical system is promoted as an independent and renewable source of energy from sustainable sanitation, environmental issues, and public health. Similar to sediment MFCs, septic tank systems must be modified to work with contemporary bio-electrochemical systems for field applications of MFCs for sanitation to be viable and practical. (Chetan Pandit, et.al.,2022)

As a result, this recently found bio-electro-chemical technique may turn out to be an effective substitute for long-term sanitation and bio-energy recovery. This bioelectric toilet (BET) technology may effectively remove nutrients, sulphur, organic matter, effluent disinfection, remove odours, efficiently degrade faecal sludge, and recover precious resources while also producing electricity to power LED lights, bio-sensors, and other electronic devices. (Chetan Pandit, et.al.,2022) MFCs’ disposal of human waste may purify toilet waste and generate electricity. In order to extract potential energy from human waste, built an eLatrine MFC using inexpensive corrugated cardboard as an electrode material and substrate being human excrement. A low open circuit voltage of 200 mV was achieved by Perlow’s initial study on the microbial oxidation of MFC waste.

A dual-chamber MFC that processed real human waste eliminated 71% of the total chemical oxygen demand (TCOD), 88.1% of the soluble COD, and 44% of the NH4 + while generating 70.8 mW/m2 of energy.

During waste-water treatment at a septic tank, three units of air cathode column MFCs were united in a stack configuration, yielding a power density of 142 6.71 mW/m2.

Recent research using septage waste-water in a 20 L MFC produced 99.8%, 86.4%, and 82.8 percent removal efficiency for total suspended solids (TSS), biochemical oxygen demand (BOD), and nitrate when working at 1000 ohms. (Leton et al.,2016 ) *Pseudomonas otitidis* AATB4 bacteria, which was isolated from septic tank effluent, had a coulombic efficiency (CE) of 15% and a current density of 800 mA/m2 in a plexiglass MFC . Thus, there have been some major attempts to extract energy from human waste, but the scaling up of such MFCs is hampered by their poor power output.

Faecal sludge is a semisolid slurry that is generated during the collection, storage, or treatment of human excreta and black water, which degrades more gradually than human excreta. The chemical energy contained in faecal sludge or activated sludge organic matter may be bio-electro-chemically converted into electrical energy using MFC. Faecal matter was used as the substrate in an experiment to attain an operating voltage of 0.45 to 0.65 V with a CE of 1.5 to 4.3 percentage. Additionally, faecal sludge’s microbial population can be used as an electro-genic inoculum, opposing the need for external seeding at MFC beginning.

The chemical energy included in urine can be recovered as electricity in MFC or as hydrogen gas during electrolysis in MECs by applying an external potential. Current improvements in microbial electrochemical systems have made MFC/MECs capable of offering a fresh method for treating urine. Recently, a urine waste-to-energy method for making bio-electricity in MFCs was reported, and work is still being done on real world applications to charge mobile phone batteries with stacked MFC units. Applying a three stage integrated MFC system, described a novel technique for recovering struvite and producing energy from human urine. Actual and synthetic urine were employed in an MFC to accomplish ammonium recovery and concurrent energy retrieval. (Kuntke et al., 2012)

Additionally tested fake urine in a 1mL MFC and found that it formed 962.94 mW/m3 of net power. In field testing of a Pee power urinal in 2016, and achieved a maximum power of 800 mW and a COD elimination efficiency of over 95% when the light was connected directly to roughly 432 MFC units. The application of MFCs for concurrent power harvesting and direct urine treatment was revealed in this first field test of an MFC system for urine waste-water. Since that time, MFC for ammonia recovery has used urine as a common substrate. (Ghangrekar MM et al.,2008)

**3.8. CHALLENGES AND FUTURE PROSPECTS**

In order to limit the losses caused by activation, ohmic, and concentration overpotentials, it becomes essential to discuss low power densities in MFC operation. Additionally, losses that are caused by reaction processes not required or microbial metabolic reactions should be enmarked as they do not benefit the process.

It is also necessary to target any microbial metabolic reactions or anodic chamber reactions that are detrimental to the process. On the other hand, it is necessary to increase system volumetric capacity while minimising internal energy losses. For avoiding catastrophic losses, stacking MFCs is a typical option.

Additionally, research on tubular and other layered techniques is also going ons. Additionally, taking action is necessary to increase the exo-electrogenic microbial population density, which appears to be constrained for reasons other than the availability of attachment positions on the electrode surface. This population density may be increased by bio-augmentation and may be influenced by potential field effects that may exist in the electrode due to its advantageous morphology and conductivity.

By modifying electrode surfaces and covering them with active catalysts, ongoing efforts are being undertaken to create improved electron transfer mechanisms between the electrode and biocatalyst.

1. **CONCLUSION**

This book chapter provided the background for use of MFC technology, substrates as waste materials used as fuel from variety of sources, designs of MFC and its applications. MFC is a considerable source of renewable energy. This sustainable approach of employing MFCs can replace the pollution caused by industrialization and over exploitation of fossil-fuels. It has various uses in power generation and waste-water treatment. This system has drawn attention due to its effective utilisation of organic wastes to generate electricity. It can be a promising alternative source for electricity generation when using biofuel in place of fossil fuel. Other than power generation and waste-water treatment MFCs can be used as bio-sensors, in the production of Bio-hydrogen, etc. As every subject has its own advantages and drawbacks, MFCs too have their share of some obstacles alongside their applications. The rate of energy production is still not feasible enough to be taken into consideration for being used on practical scale. This is because of the high costs of material used in MFC components. In future certain required changes such as modification in designs, materials of electrodes, etc should be taken in consideration to make this technology more feasible in practical approach.

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